ΈΝΑς ΜΙΚΡΟΗΛΙΟΣ ΓΙΑ ΤΟΝ ΠΡΟΔΙΟΡΙΣΜΟ ΙΣΧΝΟΣΤΟΙΧΕΙΩΝ

ΕΦΑΡΜΟΓΕς ΣΤΟ ΠΕΡΙΒΆΛΛΟΝ & ΤΗΝ ΙΑΤΡΙΚΉ

GEORGE ASIMELLIS, PH.D.



Laser ...

μια ακτίνα φωτός για την Έρευνα, την Τεχνολογία και τον Άνθρωπο

Διοργάνωση: Ινστιτούτο Θεωρητικής & Φυσικής Χημείας / ΕΙΕ

Ο ΕΝΆΛΙΘΟς…



THE SUN···· PLASMA SURFACE



Hot, ionized particles interwoven with magnetic fields

Visible radiation most intense in the yellowgreen portion of the spectrum

Photosphere (effective): 5,778 K

Corona: ~5×10⁶ K

PLASMA (ΠΛΆΣΜΑ)

•one of the four fundamental states of matter

 Like gas, plasma does not have a definite shape or a definite volume unless enclosed in a container;

•Unlike gas, under the influence of a magnetic field, it may form structures such as filaments, beams and double layers.









everyday examples of phenomena made from plasma:

Ionized

state

Lightning
electric sparks
Neon lights
plasma globe
Plasma trails





WHAT 'COLOR' IS PLASMA?

The glowing colors result from electron relaxation in excited states to lower energy states after they recombine

Light is emitted in a spectrum characteristic of the transition Atomic spectrum: The range of characteristic frequencies of electromagnetic radiation absorbed and emitted by an atom

E2

Ε,

ATOMIC EMISSION SPECTRA





ATOMIC EMISSION SPECTRA

Identification of atomic spectra: identification of atomic substance



LASER: ΚΒΑΝΤΙΚΟ ΕΡΓΑΣΤΗΡΙ ΦΩΤΟΣ



LASER: THE INITIAL IDEA



¹ Albert Einstein, "Über einen die Erzeugung und Verwandlung des Lichtes betreffenden heuristischen Gesichtspunkt", Annalen der Physik **17**, 132–148 (1905).

TRYING TO SOLVE THE PHOTOELECTRIC EFFECT





Energy is Quantized!



LASER: ΑΛΛΗΛΕΠΙΔΡΑΣΕΙΣ



Σχήμα 7-3-2: Η εκκίνηση της διαδικασίας laser.

AMPLIFICATION - PHOTON AVALANCHE



THE LASER BIRTH CERTIFICATE



THE FIRST LASER





Σχήμα 7-5-7: Το ενεργειακό διάγραμμα laser Ρουβιδίου.

The broadband optical pumping of a synthetic pink ruby crystal using a flash lamp is capable of raising a substantial fraction of the chromium ions to the upper laser level.

It consisted of a ruby crystal surrounded by a helicoidal flash tube enclosed within a polished aluminum cylindrical cavity cooled by forced air.

The ruby cylinder forms a Fabry-Perot cavity by optically polishing the ends to be parallel to within a third of a wavelength of light. Each end was coated with evaporated silver, one end was made less reflective to allow some radiation to escape as a beam.









Σχήμα 7-3-20: Μερική συμφωνία δύο κυμάτων, χρονική και χωρική.

Directionality (result of spatial coherence)





LASER PROPERTIES: SYNOPSIS





DO YOU EXPECT ME TO TALK?







LOOKING FOR A JOB

"Initially, the laser was called an invention looking for a job." Harry Stine

Laser-Matter interaction

Light absorption

We cannot change the laws of Physics

LASER-MATTER INTERACTION



Σχήμα 11-2-8: Αλληλεπιδράσεις φωτός με ιστό.

PHOTON ABSORPTION?



Σχήμα 1-4-3: Η απορρόφηση των φωτονίων εξαρτάται από την ενεργειακή δομή του υλικού στο οποίο προσπίπτει.

LASER-MATTER INTERACTION

Absorption - Fluorescence



Σχήμα 1-4-6: Μηχανισμός κβαντικής απορρόφησης και φθορισμού.

Photo-ionization



Σχήμα 1-4-8: Μηχανισμός φωτοϊονισμού.

PULSED LASER PHOTOABLATION



Σχήμα 11-2-9: Οι 'συνέπειες' της απορρόφησης από τον ιστό εξαρτώνται από το χρόνο αλληλεπίδρασης -διάρκεια παλμού- και την πυκνότητα ισχύος.

PHOTODISRUPTION



Σχήμα 11-2-11: Φωτόνια από excimer laser, με ενέργεια 6.4 eV, μπορούν να διασπάσουν τους μοριακούς δεσμούς κολλαγόνου, εισερχόμενα σε βάθος 0.25 μm.



LASER-INDUCED BREAKDOWN







Σχήμα 11-2-13: Κρατήρες από φωτοαποδόμηση με θερμική φωτοαποδόμηση με Nd:YAG και (β) με CO2, και (γ) με μη θερμική φωτοαποδόμηση με Excimer Laser





Σχήμα 11-2-7: Χάραξη λογότυπου της εταιρείας Lamda Physik πάνω σε τρίχ

LASER-INDUCED BREAKDOWN





LASER-INDUCED BREAKDOWN SPECTROSCOPY



111111111111111111111111

How LIBS WORKS



- Pulsed laser focused on to sample (0.5mm²)
- Intense optical energy ablates/excites small amount of sample
- OES techniques used to collect and analyze data
- Spectral data processed
- Results analyzed and displayed via specialized software

COMMERCIAL LIBS INSTRUMENT



No solvents or additives

Rapid sample preparation

Primary and trace elements (ppm), - high dynamic range

TRACER™ 2100 66cm H x 41cm D x 74cm W



Autosampler Stage

Low Z elements (Li, Be, B, C) Solids, liquids and/or slurries

- Multi-sample, multi-element analysis
- Automated calibration & analysis
- Ruggedized for factory & field

RECENT COMMERCIAL APPLICATIONS

- Oil Exploration & Drilling
 - ✓Chemostratigraphy
- Mining
 - ✓Ore Grading (Phosphates)
 - ✓On-line Process Control Ore Slurry
- Manufacturing QA
 - ✓ Catalyst Coatings precious Metals (Pt, Pd, Rh)
- Rapid Material Sorting
 - ✓Metal Sorting
 - ✓Ore Sorting Beneficiation

LIBS ANALYSIS ON OIL RIGS



On-rig, near real time, LIBS analysis Tracer analyzes drill cuttings within minutes of reaching the surface.

Chemostratigraphy locates the precise strata that is being drilled, allowing:

- Look ahead for fast drilling
- Accurate casing points
- b Accurate total depth
- b Directional drilling precision
- b Chemosteering

Operational in North Sea, Canada and in the Gulf Coast

CHEMOSTRATIGRAPHY – GEOCHEMICAL ROCK ANALYSIS



ICP-AES data	
Tracer data	

Tracer[™] data acquired in 3 hours ICP-AES data acquired in 3 days

ON-SITE CALIBRATION CURVES IN ROCK MATRICES



Fig. 6. Calibration curves for selected elements based on the analysis of Standard Reference Materials and In-house standards by Tracer[™]. Details of the elemental concentrations in individual stards are reported in Table 1.

ON-SITE PHOSPHATE ORE GRADING



Mine Overview



Sampling Ore

- Requirements
- •Multiple, bedded seams vary significantly in composition
- Visual distinction is difficult
- •Mill feed makeup is critical

ON-SITE PHOSPHATE ORE GRADING

LIBS Applicability •Simple requirements to operate

•Easy to use

•On-site ore analysis can easily classify ore – high, medium, low grade

•Results returned in minutes



Spectrochimica Acta Part B 61 (2006) 1253-1259

SPECTROCHIMICA ACTA PART B

Phosphate ore beneficiation via determination of phosphorus-to-silica ratios by Laser Induced Breakdown Spectroscopy

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Received 26 July 2006; accepted 5 October 2006

Abstract

We report development and application of an in-situ applicable method to determine phosphate ore rock quality based on Laser-Induced Breakdown Spectroscopy (LIBS). This is an economically viable method for real-time evaluation of ore phosphate rocks in order to separate highsilica pebbles prior to deep beneficiation. This is achieved by monitoring relative emission line intensities from key probe elements via single laser ablation shots: the ratio of the phosphorous to silica line intensities (P/Si ratio) provides a simple and reliable indicator of ore rock quality. This is a unique LIBS application where no other current analytical spectroscopic method (ICP or XRF) can be applied. Method development is discussed, and results with actual ore samples are presented.



Working in the field



Very Simple Environment

ONLINE CU SLURRY MEASUREMENT



ON-LINE SLURRY CU SLURRY ANALYSIS



On-line data showing dynamic response of real-time LIBS data vs. the less responsive XRF analysis.

CATALYST COATING QA ANALYSIS – SPECTRAL LINE WINDOW



CATALYST COATING QA ANALYSIS -SPECTRAL LINE WINDOW



Platinum

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ABSTRACT

ARTICLE INFO

Article history: Received 14 April 2008 Accepted 17 September 2008 Available online 9 October 2008

Keywords: Laser-induced breakdown spe

Development and application of an in-situ applicable method to provide rapid determination group metals (platinum, palladium, and rhodium) elemental concentration in automobile cata reported. Application is based on laser-induced breakdown spectroscopy (LIBS). Actual automo slurry in powder form was used to develop the application. With a method requiring approxima of examination per sample, calibration curves are presented with linear regression coefficients (and stability better than 3.0%.

RAPID METAL / ALLOY SORTING



Highly reliable and fast results

Spectra obtained and compared in 100 msec

RAPID ORE SORTING - BENEFICIATION



Method Development

Calibration with SRM

Spectra ratio identifies high Si content with one shot

RAPID ORE SORTING - BENEFICIATION



Method Development

Calibration with SRM

Spectra ratio identifies high Si content with one shot

LIBS FOR PLATED LAYER THICKNESS IDENTIFICATION & MEASUREMENT

Single spectral window (one shot) identifies all elements of

Rapid, automated measurement of layer thicknesses on steel coin blanks using laser-induced-breakdown spectroscopy depth profiling

George Asimellis, Aggelos Giannoudakos, and Michael Kompitsas

We report application of a post-real-time method to determine layer thickness on electroplated coil blanks. The method was developed on a simple laser-induced-breakdown spectroscopy (LIBS) arrange ment by monitoring relative emission-line intensities from key probe elements via successive lase ablation shots. This is a unique LIBS application where no other current spectroscopic method (inductively coupled plasma or x-ray fluorescence) can be applied effectively. Method development is discussed, and results with precalibrated coins are presented. \square 2007 Optical Society of America OCIS codes: 120.0120, 120.6200, 140.3440, 300.6360, 350.3390

1. Introduction

The Royal Canadian Mint (RCM) plating facility in Winnipeg, Canada, has identified a problem with performing quality control of their electroplated coin blanks. Typically, coins, of a ferritic steel core, are electroplated with three layers, two layers of approximately 4–8 μ m thick nickel, separated with a middle laver of 7-10 µm thick copper. A rapid and automated technique was sought to perform near-realtime quality assurance on production coin blanks to replace current methods of coin sectioning and subsequent optical metallographic measurements, because these mechanical methods are slow and labor intensive

Laser-induced-breakdown spectroscopy (LIBS) enables multielemental identification and quantitative analysis requiring little or no sample preparation. A

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schematic of LIBS arrangement is shown in Fig. 1. A high-power laser pulse is focused just above or below the sample surface, creating a localized area of material removal (less than 1 mm wide and submicrometer deep per shot). Particles from the ablated material are subsequently ionized, and thus a hot plasma is created. Upon plasma cooling, ions recombine, and consequent excited atoms relax, radiating atomic emission lines, characteristic of the ablate sample constituent elements. By spectrally examin ing the optical emission, it is possible to identify the ablated area's atomic composition. This is achieved by a high-resolution spectrograph, on which the op tical plasma radiation is directed, by means of fused silica optics and steering mirrors. The spectrum is recorded on a gated, intensified CCD detector. Thus specific quantitative element analysis can be accomplished in near real time.1-3 This work focuses on the applicability of LIBS for

rapid identification of plated layer thickness. The elemental composition of the ablated layer can be analyzed for every successive shot as the laser penetrates deeper into the material at a repeatable rate Thus the number of shots required to penetrate a specific layer can be determined by monitoring the appearance of new probe elemental lines or the sharp decline or increase in the line intensities (photon count) of existing elemental lines. Thus LIBS can provide detailed knowledge of the layered depth pro file on a very localized sample area, less than 1 mm wide.4-13 It is this characteristic depth-profiling capability of LIBS that provided motivation for thi work

20 February 2007 / Vol. 46, No. 6 / APPLIED OPTICS



interest

8 4

LIBS FOR PLATED LAYER THICKNESS IDENTIFICATION & MEASUREMENT

Successive Ablation Shots Penetrate through layers.

When interface is reached, change in amplitude and/ or slope is



HALOGEN DETECTION

EISEVIER	Superstructuring Auto Dart D. 60 (2006) 1122 1120	SPECTROCHIMICA ACTA Part B	
LESEVIER	Spectrochimica Acta Part B 60 (2005) 1152 – 1159	www.elsevier.com/locate/sab	
Controll fluor	ed inert gas environment for enhanced ine detection in the visible and near-inf laser-induced breakdown spectroscopy	chlorine and frared by y^{x}	
George Asimell	is ^{a,*} , Stephen Hamilton ^b , Aggelos Giannoudakos ^c ,	Michael Kompitsas ^d	
^a Aristotle Uni ^b Anda ^d National Hellenic Research Fa	versity of Thessaloniki, Department of Physics, Solid State Section, 313-1, GR 5412 or Technology, 7 Millemium Way Springvale Business Park, Belfast, BT12 7.41, No Chemical Engineering Department, National Technical University of Athens, Athen sundation-NHRF, Theoretical and Physical Chemistry Institute-TPCI, 48 Vas. Konst	4 Thessaloniki, Greece rthern Ireland s, Greece antinou Ave., GR 11635 Athens, Greece	
	Received 30 November 2005; accepted 30 May 2005 Available online 23 June 2005		
Abstract			1.1.1.1

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SPECIAL ISSUE PAPER

George Asimellis · Aggelos Giannoudakos Michael Kompitsas

New near-infrared LIBS detection technique for sulfur

Received: 11 January 2006 / Revised: 3 February 2006 / Accepted: 3 February 2006 / Published online: 11 March 2006 © Springer-Verlag 2006

Abstract Sulfur has been detected in a spectral window (around 868 nm) previously unexplored by laser-induced breakdown spectrometry (LIBS), using an ablation laser with an ultraviolet wavelength, a gated detector, and inert ambient gas at a low, controlled pressure. This spectral window enables new-generation gated iCCD cameras to be weapons in non-standard environments [8, 15, 16]. used, which have adequate quantum efficiencies up to 900 nm. Application of our technique can substantially thoroughly investigated by LIBS can be attributed to a improve signal strength and thus extends the ability of number of issues related to their energy level distributions: LIBS to detect many nonmetallic elements.

Keywords Halogens · Ambient gas · Plasma · Sulfur

compounds in consumer electronics [5,14] and chemical The fact that nonmetallic elements have not been

products [10], classification of minerals and drill core samples in oil exploration [11], on-line analysis in steel

processing [7], detection of air-polluting hazardous gases

[12] and aqueous solutions [13], determination of organic

their strongest emission lines lie in the vacuum ultraviolet (VUV) range (125-190 nm). Practical application of LIBS in this VUV range is complicated by strong absorption from atmospheric oxygen (which necessitates that the light travels through a vacuum, or at least an oxygen-free atmosphere), the lack of gated (iCCD) detectors with adequate sensitivity, and the need to use non-UV-absorbing

ELSEVIER

Introduction

Spectrochimica Acta Part B 61 (2006) 1270-1278

SPECTROCHIMICA ACTA PART B

www.elsevier.com/locate/sab

Near-IR bromine Laser Induced Breakdown Spectroscopy detection and ambient gas effects on emission line asymmetric Stark broadening and shift

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Abstract

Investigation via Laser-Induced Breakdown Spectroscopy of a near-infrared bromine emission line (827.2 nm) with an UltraViolet ablation laser and a gated detector is reported. The effects of ambient pressure and gas species (air, O2, N2 and He) on the atomic emission line strength and spectral profile were systematically investigated. Substantially improved signal strength and reduced background radiation are demonstrated near 100 mbar ambient pressure with all gases. Optimal results were achieved when helium was used. Asymmetric broadening and shift of the 827.2 nm bromine line, attributed to pressure-dependent Stark effect has been revealed. This effect is prominent when air, oxygen or nitrogen are present and is much less manifested when helium is used. Possible interpretations of this effect are presented. © 2006 Published by Elsevier B.V.

Keywords: Laser-induced breakdown spectroscopy; Halogen detection; Bromine content; Stark effect; Ambient gas effects

Efficient quantitative detection for halogens is necessary in a wide range of applications, ranging from pharmaceutical products to air polluting hazardous gases or organic compounds used as chemical weapons. Detection of the non-metallic elements such as fluorine (F) and chlorine (Cl) presents particular difficulty, because strong emission lines originating from their resonance states lie in the VUV spectral range (110-190 nm). In the present work we detect F and Cl in the upper visible and in the near IR (650-850 nm) under controlled inert gas ambient atmosphere. Investigation of the controlled atmosphere effects suggests that there exists an optimum pressure range that optimizes signal strength and quality. Ablation and ionization were achieved with a UV laser at 355 nm, and a gated GaAs photocathode-based detector was used for detection with quantum efficiency in the range of 20% in the wavelengths of interest. Our results indicate that our approach provides quantitative detection with linearity over at least two orders of magnitude that is achieved without the need for Internal Standardization Method, and improved limits of detection. In particular, fluorine has been detected for concentration values down to 0.03 wt.% Definite spectral assignment revealing all major emission lines centered around 837 nm for F and 687 nm for Cl has been obtained for the first time in Laserinduced breakdown spectroscopy application.

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LASER & OPHTHALMOLOGY

Cornea (refractive

surgery)

Retina (photocoangulation) Laser interaction with the eye!

Iris (open angle glaucoma) Laser trabeculoplasty (LTP)

Crystalline Lens (laser cataract surgery)

OPHTHALMIC LASERS

Ophthalmic lasers provide good examples of three fundamental laser applications, based on: total output energy (thermal effects, photocoagulation)

output power (tissue ionization, photodisruption)

photon energy (breaking molecular bonds, photoablation)

spectral transmission properties of the eye

LASER WITHIN THE EYE!



Absorption by Ocular Media



Σχήμα 5-3-15: Φασματική διαπερατότητα του οφθαλμού.

ABSORPTION SPECTRA OF IMPORTANT OCULAR CHROMOPHORES



PHOTODISRUPTION



because atoms typically must absorb > 10 eV to ionize need "multi"photon absorption for ionization

POSTERIOR CAPSULOTOMY

to open up an opacified lens capsule resulting from prior extracapsular cataract surgery









PHOTOABLATION BY EXCIMER LASER

Removal of corneal tissue by high energy photon directly breaking molecular bonds with sufficient

photon energy,

193 nm, 6.4 eV energy

Excimer laser





THE LASIK PROCEDURE



CREATING A 'FLAP' MECHANICALLY







CREATING A 'FLAP' WITH A LASER







FLAP CREATION BY MICROABLATIONS

canal line separa 2.5 µm

gas bubbles

spot separation

8.0 µm

Nd:YAG pulsed laser Pulse duration: few 10⁻¹⁵ sec Femto-second laser



Needs different wavelength!



Σχήμα 5-3-15: Φασματική διαπερατότητα του οφθαλμού.

FEMTO-SECOND ABLATION WITHIN THE CORNEA



ABLATION WITHIN THE CORNEA



LIFTING OF THE FLAP, EXCIMER ABLATION



CATARACT REMOVAL: THE MOST COMMON OPHTHALMIC SURGERY









CATARACT SURGERY









CATARACT SURGERY WITH FEMTOSECOND LASER







